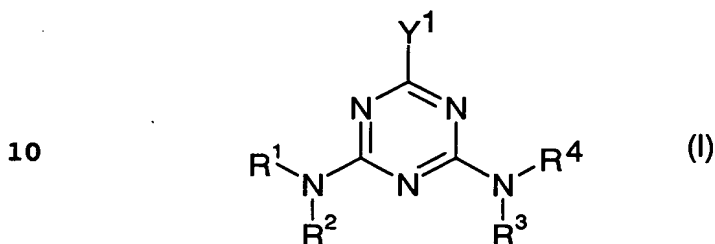


We claim:

1. A process for preparing alkoxy carbonylaminotriazines of the
5 formula I



15 in which

Y¹ is hydrogen, C₁-C₄-alkyl, phenyl optionally substituted
by C₁-C₄-alkyl, C₁-C₄-alkoxy or halogen, or a radical of
the formula NR⁵R⁶ and

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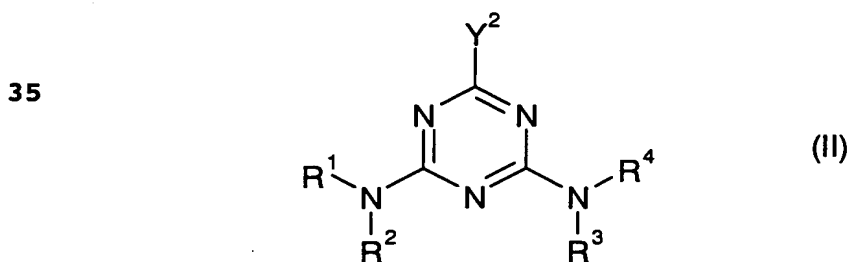
R¹, R², R³, R⁴, R⁵ and R⁶ are each independently hydrogen or a
radical of the formula COOX or X where X is C₁-C₁₃-alkyl
whose carbon framework may be interrupted by 1 or 2
oxygen atoms in an ether function and/or be substituted
25 by hydroxyl, or C₃-C₆-alkenyl,

25

with the proviso that at least one of the radicals R¹ to R⁴ in
formula I or, when Y¹ is NR⁵R⁶, at least one of the radicals
R¹ to R⁶ is COOX,

30

by reacting a triazine of the formula II



40

in which

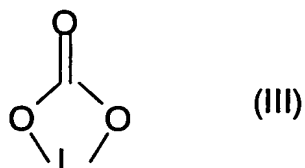
Y² is hydrogen, C₁-C₄-alkyl, amino or phenyl optionally
45 substituted by C₁-C₄-alkyl, C₁-C₄-alkoxy or halogen, and

13

R¹ to R⁴ are each as defined above,

with the proviso that, in formula II, when Y² is not amino,
at least one of the radicals R¹ to R⁴ is hydrogen,

with carbonic esters in the presence of an alcohol and of a
base, which comprises reacting the triazine of the formula II
with a cyclic carbonic ester of the formula III



15 in which

L is ethylene, 1,2- or 1,3-propylene, or 1,2-, 1,4-, 2,3-,
or 1,3-butylene,

and also optionally with minor amounts of an acyclic carbonic
ester of the formula IV



25 in which

Z¹ and Z² are each independently C₁-C₈-alkyl,

30 and a C₁-C₁₃-alkanol whose carbon framework may be interrupted
by 1 or 2 oxygen atoms in an ether function and/or be
substituted by hydroxyl, or a C₃-C₆-alkenol, in the presence
of an alkali metal alkoxide or alkaline earth metal alkoxide
as a base.

35 2. A process as claimed in claim 1, wherein a C₁-C₁₃-alkanol is
used.

40 3. A process as claimed in claim 1, wherein the base used is an
alkali metal alkoxide.

4. A process as claimed in claim 1, wherein a cyclic carbonic
ester of the formula III in which L is ethylene or
1,2-propylene is used.

5. A process as claimed in claim 1, wherein the reaction is carried out at a temperature of from 20 to 180°C.
6. A process as claimed in claim 1, wherein the reaction is carried out with from 1 to 50 mol of alkanol, based in each case on one molar equivalent of amino groups in the triazine of the formula II.
7. A process as claimed in claim 1, wherein the reaction is carried out with from 0.1 to 10 mol of cyclic carbonic ester, based in each case on one molar equivalent of amino groups in the triazine of the formula II.
8. A process as claimed in claim 1, wherein the reaction is carried out with from 0.1 to 10 molar equivalents of alkali metal alkoxide or alkaline earth metal alkoxide, based in each case on one molar equivalent of amino groups in the triazine of the formula II.
9. A process as claimed in claim 1, wherein triazine II and alkanol are initially charged and then, in any desired sequence, alkali metal alkoxide or alkaline earth metal alkoxide, in the solid state and/or dissolved in alkanol, and carbonic ester are metered in, and alkali metal alkoxide or alkaline earth metal alkoxide and carbonic ester can be metered in fully before the commencement of the reaction or partly before the commencement of the reaction and partly after the commencement of the reaction.
10. A process as claimed in claim 1, wherein from 0 to 25 mol% of the cyclic carbonic esters of the formula III may be replaced by acyclic carbonic esters of the formula IV.